Mechanics of Diffusion-Induced Fractures in Lithium-ion Battery Materials

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ABSTRACT

Our study is motivated by the need for development and deployment of reliable and efficient energy storage devices, such as lithium-ion batteries. However, the rate-capacity loss is the key obstacle faced by current lithium-ion battery technology, hindering many potential large-scale engineering applications, such as future transportation modalities, grid stabilization and storage systems for renewable energy. During electrochemical processes, diffusion-induced stress is an important factor causing electrode material capacity loss and failure. In this study, we present models that are capable for describing diffusion mechanisms and stress formation in LiFePO₄ nanoparticles, a lithium-ion battery cathode material which promises an alternative, with the potential for reduced cost and improved safety. To evaluate mechanics of diffusion-induced fractures, a plate-like model is adopted with anisotropic materials properties and volume misfits during the phase transformation are considered. Stress distribution at phase boundaries and fracture mechanics information (energy release rates and stress intensity factors) are provided to further understand the stress development due to lithium-ion diffusion during discharging. This study contributes to the fundamental understanding of kinetics of materials in lithium-ion batteries, and results from our stress analysis provides better electrode materials design rules for future lithium-ion batteries.

INTRODUCTION

Lithium ion batteries have become a widely known commodity for satisfying the world’s mobile energy storage needs. Applications for these high power batteries include anything from laptop computers and mp3 players to orbiting satellites and electric tools. The advantages of Li-ion over other types or rechargeable batteries such as Nickel-Cadmium and Metal Hydride types lie in their superior energy-to-weight ratio, quicker recharge times, and increased cycle life. In addition, as the world becomes more concerned about rising oil prices and CO₂ emissions, the need to increase hybrid or electric vehicle production and performance (along with other high power applications) has fueled research to improve Li-ion battery performance and even discover new battery materials.

Since the conception of Lithium-Iron-Phosphate (LiFePO₄) as a possible cathode for today’s Li-ion batteries [1], this material with olivine structure has been a focal point for such experimentation and discussion in the Li-ion battery realm. LiFePO₄ boast several attractive qualities including a relatively high theoretical capacity of 170 mAh/g, great structure stability, long cycle life, impressive safety attributes and environmentally benign qualities. It’s also widely available across the world at a relatively inexpensive price. But with the pros, cons also exist such as extremely low electrochemical conductivity for raw LiFePO₄ and a tendency for

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capacitance loss over its cycle life. Nevertheless, the favorable characteristics of LiFePO$_4$ [2-4], especially in safety, provide a strong base that research advancements can build on to make LiFePO$_4$ one of the most competitive cathode material’s in today’s growing markets for battery powered electronics, power tools and hybrid electric vehicles to name a few.

Figure 1. A two-phase LiFePO$_4$ model, in which lithium-ion diffuse along the b-axis and the phase boundary moves along the a-axis [5].

Computational and experimental observations have found that lithium-ion diffusion in LiFePO$_4$ is one-dimensional confined along the b-axis [6-10]. Delmas et al. [11] have proposed a “domino-cascade” model to describe the phase boundary movement along the a-axis for LiFePO$_4$, when lithium-ion diffuse along the b-axis during charging/discharging. In the “domino-cascade” model, the interface is limited to one block, therefore, either a fully lithiated or an unlithiated phase exists in a partially discharged LiFePO$_4$ particle. That is, a diffused and transit interfacial zone was not considered in the domino-cascade model. Moreover, other experiment results reveal the coexistence of FePO$_4$ and LiFePO$_4$ phases in a single particle, where stripe-like juxtaposed phase boundaries are observed [12-15]. A co-existing two-phase system under fast discharging was also computationally predicted by Cogswell and Bazant, Bai et al. [16], and Van der Ven et al. [17].

To better understand the stress evolution with the phase boundary propagating along the a-axis [12-15], we have previously focused on studying the stress distribution on the phase boundary between LiFePO$_4$ and FePO$_4$ phases (Figure 1) [5]. We used a thermal stress analysis approach via a finite element method to obtain diffusion-induced stresses in a plate-like LiFePO$_4$ particle. The particle is divided into 10 layers along the a-axis and each layer contains one Li-ion diffusion channel. That is, each layer is corresponding with 10% nonstoichiometric lithium-ion composition (Figure 1b).

Figure 3. Normalized $\sigma_{xz}$ evolution of a single LiFePO$_4$ (LFP) particle. Normalized $\sigma_{xz}$ exhibits a non-negligible value during the phase transformation and it might play an important role on fracture mechanics and crack propagation.

LiFePO$_4$ phase has a larger molar volume and FePO$_4$ phase has a smaller molar volume, where a coherent interface exists between these two phases during the phase transformation [18].

Figure 4. A representative bi-materials LiFePO$_4$ particle. An initial flaw exists between two phases.
As a result, the LiFePO₄ phase would be under compression and the FePO₄ phase would be under tension due to the constrains of the coherent interface. The results from our finite element analysis indicate that the particle is under compression before 70% lithiation for normalized $\sigma_{xx}$ (Figure 2). Our finite element simulation also revealed that $\sigma_{xz}$ exhibits a non-negligible value during the phase transformation (Figure 3), and it is suggested that $\sigma_{xz}$ might play an important role on fracture mechanics and crack propagation in electrode particles [19]. In the current study, we plan to unravel effects of diffusion-induced stresses on lithium-ion battery materials. In particular, to study fracture mechanics of nano-particles with pre-existing flaws.

MATERIALS AND METHOD

To further study fracture mechanics and crack propagation in LiFePO₄ particles, virtual crack closure technique (VCCT) with the ANSYS finite element software (ANSYS, Inc. Canonsburg, Pennsylvania) is utilized to obtain energy release rates and the stress intensity factors [20]. Energy release rates are used to predict the fracture tendency and the stress intensity factors are used to relate stress fields around crack tips. The fracture mechanics information (energy release rates and stress intensity factors) are calculated from the command-coding interface of ANSYS [21], and they are currently not available on the graphical user interface of ANSYS.

Two-phase plate-like LiFePO₄ finite element models are generated and orthotropic material properties are adopted for both phases [22]. An interfacial crack is considered to run parallel to the $bc$-plane (along the $c$-axis) where phase boundaries are present (Figure 4) [23]. Three different sizes of finite element model are generated and the corresponding numbers of unit cells are as follows: (a) 500-nm × 300 nm × 225-nm with 125,000 unit cells, (b) 200-nm × 120 nm × 90-nm with 8,000 unit cells, and (c) 100-nm × 60 nm × 45-nm with 1,000 unit cells, respectively. The numbers of the unit cell are approximated based on the LiFePO₄ lattice constants: $a = 10.3$ Å, $b = 6.0$ Å, and $c = 4.7$ Å [24]. Assuming displacements along the $a$- and $c$-axes are independent of $b$-axis and there are no normal tractions on the $ac$-plane, the plane stress assumption is applied. A total of 21 finite element simulations are conducted, whereas the initial crack is considered from the top face with variable lengths of cracks sizes 0.05 – 0.8 L/d, in which L represents the crack length and d represents the particle size along the $c$-axis (Figure 4). The crack opening is set as 0.5 nm in the $a$-direction to stay proportional to experimentally observed cracks [23]. Displacements were applied to the LiFePO₄ phase according to the misfit strains previously measured and reported in the literature: expansion occurs along the $a$-direction ($\varepsilon_a = 5.03\%$) and extraction occurs along the $c$-direction ($\varepsilon_c = -1.9\%$) [25] (Figure 4). Two-dimensional quadrilateral elements are used in the finite element analysis, and such eight-node parabolic elements allow for more flexibility and improved accuracy in contrast to four-node elements [26, 27].

A crack will propagate if the total energy release rate of Modes 1 and Mode 2, $G_T$, is larger than approximately twice the surface energy of the particle ($G_T > 2\gamma$) [28]. A first-principle analysis by Wang et al. has reported a $\gamma$ value of 0.66 N/m for LiFePO₄ in the (100) crack face orientation [29]. In the current study, we collect finite element results satisfying $G_T > 2\gamma$, and the difference between these two energies is used to predict crack growth patterns. That is, crack propagation is eminent and a crack will grow until the surface energy from the newly formed crack faces brings the particle back to equilibrium with the strain energy release rate. First, we calculate the additional energy release rate for a 0.01-nm crack extension for all models, $E_R$. 
Second, we calculate the additional surface energy for a 0.01-nm crack extension for all models, \(E_S = 0.01 \times b \times \gamma\), where \(b\) is the particle size in the \(b\)-axis. Last, we calculate the required crack extension \(d_a\) via the ratio of \(E_R\) and \(E_S\).

**RESULTS AND DISCUSSION**

The results of the finite element method based-VCCT indicate that Mode I fracture is dominant for all models. **Figure 5** reveals that \(G_I\) is highly dependent on the crack size for all three different particle sizes. Critical points occur at around \(L/d = 0.5\) and 0.6, where energy release rates increase with the increased crack lengths, and then energy release rates begin to level off. In particular, smaller particles (100 nm \(\times\) 60 nm \(\times\) 45 nm) are able to better accommodate initial flaws with the crack propagation when \(L/d \leq 0.2\). Although the crack propagation is possible in other planes, literature and experimentations have reported cracks are observed in the \(bc\)-plane between phases while propagating in the \(c\)-axis [13]. Less energy release rates are observed in the Mode II fracture, however, these values are considered to contribute to the total energy release rate \(G_T\) for the crack propagation, and it is discussed in the following section.

**Figure 5.** Energy release rates for two modes for three different particle sizes.

**Table 1** shows the fracture mechanics and crack propagation results from the finite element method-based VCCT simulation. It is observed that Mode I stress intensity factor increases with larger crack to particle size \((L/d)\). Moreover, when \(G_T\) is compared with \(2\gamma\), the energy difference \(\Delta E = G_T - 2\gamma\) is shown in **Table 1**, and it is used to calculate the required additional strain energy release rate for a crack extension. It is observed that crack propagation is highly dependent on the initial crack size and particle sizes.
<table>
<thead>
<tr>
<th>L/d</th>
<th>K (MPa-m$^{1/2}$)</th>
<th>G (N/m)</th>
<th>2γ (N/m)</th>
<th>ΔE (N/m)</th>
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**CONCLUSIONS**

In summary, mechanics of diffusion-induced fractures of lithium-ion battery materials is presented. Our preliminary results showed that mechanical stresses on the phase boundary would likely promote crack propagation (Figure 3), if initial flaws exist inside electrode nano-particles. Finite element method based-VCCT is adopted in the current study to obtain strain energy release rates and stress intensity factors. Results show that energy release rates (Figure 5) and stress intensity factors are highly related to the initial flaw sizes and particle sizes (Table 1). In particular, smaller particles (100 nm × 60 nm × 45 nm) are better to accommodate initial cracks. This is a first approach towards utilizing stress fields on the phase boundary to better understand fracture mechanics of lithium-ion battery electrode materials. The results of the current study helps improve battery structural design by providing a better understanding of micromechanics and a base for future fatigue analyses, which may incorporate a variety of electrode materials and additional fatigue life parameters. Studies have shown that the application of nano-layer coatings to increase electronic conductivity along with reducing particle size to effectively reduce the diffusion path and decrease internal stress within the particle [30, 31]. This research will hopefully elucidate a relationship between micromechanics and battery usage to help in the design of a higher performance, longer lasting battery for the future.
REFERENCES